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fundamentals of general, organic, and biological chemistry second edition john mcmurry cornell university mary e. castellion norwalk, connecticut m prentice hall, upper saddle river, new jersey 07458 . brief contents applications and interludes xv preface xvii a note to the student xxv **test bank for fundamentals of general organic and ...** - 32) explain why it is unlikely that an organic molecule would have an odd number of valence electrons. answer: in most molecules, the lewis dot structure includes bonds that consist of one, two, or three pairs of electrons. the remaining electrons are drawn as lone pairs on one or more atoms. note that both of these include pairs of electrons. also **fundamentals of general organic and biological chemistry ...** - complete test bank for fundamentals of general organic and biological chemistry 7th edition by mcmurry, ballantine, hoeger, peterson download sample. test bank for fundamentals of general organic and biological chemistry 7th edition mcmurry - 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